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A Novel Double Friedel-Crafts Reaction: A New Entry into Bicyclo[3.2.2]nonane system.

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Abstract: A novel synthesis of dibenzobicyclo[3.2.2]nonane systems is described through a double Friedel-Crafts reaction.

Recently we reported¹ a single step synthesis of 1-aryltetralins from 1-tetralols and phenol by the Friedel-Crafts reaction. During the preparation of a potential intermediate 7 required for building the A-ring aromatic steroids² from the tetralol **6** we unexpectedly encountered the formation of a bicyclo[3.2.2]nonane derivative **10**. The natural sesquiterpenes³ rotundene & rotundenol, the polycyclic toxin CBT⁴ and Xanthoquinidins⁵ possess the unique bicyclo[3.2.2]nonane nucleus. In view of the importance of the bicyclo[3.2.2]nonane skeleton in bicycloaromaticity⁶, coupled with the difficulties involved with their preparation⁷, prompted us to report our results in this paper.

Reaction of the diol $\mathbf{6}$ with phenol in the presence of anhydrous aluminium chloride in 1,2-dichloroethane gave the dibenzobicyclo[3.2.2]nonane derivative 10 in 70% yield.



The diol $\underline{6}$ was prepared from the alcohol⁸ $\underline{1}$ through the sequence of reactions indicated below. Thus the alcohol $\underline{1}$ was oxidized to the aldehyde $\underline{2}$, which was methylated to give the product $\underline{3}$. Reduction of $\underline{3}$ with sodium borohydride followed by acetylation afforded the acetate $\underline{4}$, which was oxidized with PDC-BuOOH to yield the ketone $\underline{5}$, which gave a diastereometric mixture of diols $\underline{6}$ with sodium borohydride.

The mechanism of formation of 10 from 6 appears to involve the initial formation of the aryltetralin derivative 8 which undergoes an intramolecular Friedel-Crafts reaction probably through the dienone 2 to the compound 10. This mechanism is supported by the fact that while the aryl tetralin 8, synthesized separately, is smoothly transformed into the bicyclo[3.2.2]nonane derivative 10 in the presence of anhydrous AlCl, in 1,2-dichloroethane, the isomeric compound 7 is recovered unchanged under similar conditions. The exclusive formation of 8 in preference to 7 in this reaction, can be attributed to the steric hindrance exhibited by the

methyl group to the incoming aryl group, thus disfavouring the formation of $\underline{7}$. The hydroxymethyl group occupies the pseudoequatorial position.

Lithium/liq.ammonia/BuOH reduction of the methyl ether of 10 followed by hydrolysis with acid gave the unsaturated ketone 11, thus confirming the structure of 10.

To check the generality of the reaction, the diol $\underline{6}$ was reacted with different cresols and anisole, and in all these cases the corresponding bicyclo[3.2.2]nonane derivatives⁹ were obtained in good yields as indicated in Table-1. Even the diol <u>17</u> synthesized from the alcohol <u>12⁸</u>, through the intermediates <u>13</u> to <u>16</u>, employing the same sequence of reactions as in the case of the diol <u>6</u>, underwent smooth double Friedel-Crafts reaction yielding the bicyclo[3.2.2]nonane <u>21</u>, thus confirming the generality of the reaction (Table-1).

OCOCH3



Table – 1.



* Yields refer to pure isolated yields

In summary, we describe a simple and efficient method of synthesis of a bicyclo[3.2.2]nonane skeleton by a one pot double Friedel-Crafts reaction.

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- 9. ¹H NMR(90MHz,CDCl₃) data for selected compounds:
 - **10.** δ 1.45(s,3H,CH₃), 1.5-2.4(m,4H,CH₂), 2.91(d,J=3.64Hz,2H,ArCH₂), 3.76(s,3H,OCH₃), 4.65(s,1H,OH), 6.48-7.1(m,6H,ArH).
 - **18.** δ 1.48(s,3H,CH₃), 1.4-2.4(m,4H,CH₂), 2.21(s,3H,ArCH₃), 2.93(d,J=5.4Hz,2H,ArCH₂), 3.8(s,3H,OCH₃), 4.6(bs,1H,OH), 6.56-7.0(m,5H,ArH).
 - **19.** δ 1.52(s,3H,CH₃), 1.3-2.4(m,4H,CH₂), 2.34(s,3H,ArCH₃), 2.93(d,J=5.4Hz,2H,ArCH₂), 3.84(s,3H,OCH₃), 4.68(bs,1H,OH), 6.4-7.0(m,5H,ArH).
 - **<u>20.</u>** δ 1.48(s,3H,CH₃), 1.4-2.4(m,4H,CH₂), 2.92(d,J=3.6Hz,2H,ArCH₂), 3.76(s,3H,OCH₃), 3.78(s,3H,OCH₃), 6.5-7.2(m,6H,ArH).
 - **<u>21.</u>** δ 1.44(d,J=1Hz,3H,CH₃), 1.3-2.3(m,4H,CH₂), 2.92(d,J=3.6Hz,2H,ArCH₂), 3.7(s,3H,OCH₃), 4.6(bs,1H,OH), 6.4-7.16(m,6H,ArH).

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